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The solution is saturated. b. All 15 g of  $KClO_3$  will dissolve; the solution is unsaturated. 3. a. 250. g b. 50. g c. KI 4. a. 1.027 kJ/g b. The temperature of the system will increase because the enthalpy of solution is negative, indicating that the reaction is exothermic, giving off energy as heat and warming up the system. SECTION 3 SHORT ...

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phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

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